Effect of Heavy Metal (Lead) on Human Health

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Abstract

The pollution sources of heavy metals Pb, Fe, Cr, Zn, Cd, Ni, Cu and Hg increase their concentration both at the global as well as local level. The increase in concentration of metals in a localized area reach sometime to a significantly higher level and be one deleterious to biological population and human beings. Several disasters of the metal poisoning have been recorded from time to time which have been caused great ecological damage and led to a large number of human causalities. The recommended maximum lead concentration in drinking water is 0.01 ppm.

Key words : pollution, Heavy Metals.

Most of the elements present in the "periodic table" are metals^{6,9}. The metals work as electron pair acceptors and react with electron donor to form a variety of compounds like coordination compounds, metal complexes or donor acceptor compounds, chemically, the metals have been classified variously, but biologically^{3,5,7,10}, we used the term heavy metals to denote certain category of metals which are most deleterious ecologically. These heavy metals have a specific gravity of 4 - 5 times greater than that of water and usually belong to atomic number 22 - 34 and 40 - 52 and members of lanthanider and actinides of the periodic table^{7,8,9,11,12}.

Physico-chemical analysis of water samples of Bhopal city was carried out by standard methods of APHA(2005)¹. Lead can be determined by following methods-100 ml acidified water sample + 20 ml HNO_{3} + 50 ml ammonium citrate cyanide solution in 25 ml separating funnel. Mix and cool it at room temperature add 10 ml dithizone .Shake funnel and separate organic layer cherry red in colour measure absorbance at 510 nm.

Environmental pollution is the result of urban industrial technological revolution and speedy exploitation of every bit of natural resources rapid industrialization have created adverse effect on all living organism in the biosphere .Toxic metals are added in aquatic system from industrial process, domestic sewage discharge, street dust, land sun of and fossil fuel burning. Traces of heavy metals such as Hg, Cd, Pb, As, Co, Mn, Fe and Cr have been identified as deleterious to aquatic system and human health waste containing high concentration of toxic metals either separately or in combination are extremely toxic to all organism. Lead compounds discharge in auto exhaust main lead compounds are-

Cigarette smoking may be an important source of lead to man. An average cigarette contains about 0.8 gram of lead. A heavy smoker can inhale a quatity of lead as higher 4.8 gram per day.

Toxic response of heavy metals to man:

Anemia and disruption of hemoglobin synthesis, damage to nervous system and kidneys, brain damage.

The major biochemical effects to Pb has been attributed to its interfearence with heme synthesis which gives rise to hematological damage. Pb is reported to inhibit several of the key enzymes which are involved in the overall process of heme synthesis where by the metabolic intermediates get accommodated one such intermediate has been delta-amino levulinic acid. An important phase of heme synthesis involves the conversion of deltaaminolevulinic acid to porphovilinoges:

$$\begin{array}{c} O\\ ||\\ HOOC - CH_2 - CH_2 - C - C - C - COOH\\ |\\ NH_3\\ Delta aminolevulinic acid\\ (a)\\ (ALA dephydrase : cytoplasm)\end{array}$$

$$\begin{array}{c} HOOC - CH_2 - C - C - CH_2 - CH_2 - COOH \\ I & I \\ C & C \\ H_2N - CH_2 & N \\ I \\ H \\ Porbphobilinogen \\ (b) \end{array}$$

As, Pb inhibits the ALA – dehydrase enzyme it cannot react further to form (b) porbphobilinogen. The overall effects involves the disruption of the synthesis to hemoglobin as well as the other respiratory pigments like cytochromes, which need heme. Finally Pb does not allow the using of O_2 and glucose for life sustaining energy production. At level of Pb in blood(>0.8 ppm) symptoms of anemia due to the deficiency of hemoglobin appears. Elevated Pb levels more than (>0.5-0.8 ppm) in the blood may bring about kidney disfunction and finally cause brain damage.

Methods for removal of lead from the body all depends to some extent, on coordination of the metal to give a soluble species that may be excreted from the body. The most common methods used today is chelation using CaNa2 EDTA. Other approaches to administer a low calcium high phosphate diet are to prescribe sodium citrate and 2,3 dimercapropanol.

 $2HSCH_2 - CH - CH_2OH + Pb^{2+}$

$$SH$$

$$\downarrow$$

$$H_2 C - S$$

$$\downarrow$$

$$H_C - S$$

$$\downarrow$$

$$HC - S$$

$$+4H$$

$$\downarrow$$

$$CH_2CH$$

The recommended maximum lead concentration in drinking water is 0.10 ppm.

The lead concentration in upper lake of Bhopal city is found to be 0.086 ppm. Toxicity of lead is more acute in infants and children than the adults.

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